# organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

# Silvia Russi,<sup>a</sup>\* Leopoldo Suescun,<sup>a</sup> Mirta Mischne<sup>b</sup> and Alvaro W. Mombrú<sup>a</sup>

<sup>a</sup>Laboratorio de Cristalografía y Química del Estado Sólido, Facultad de Química, Universidad de la República, Av. Gral. Flores 2124, Casilla de Correo 1157, Montevideo, Uruguay, and <sup>b</sup>Instituto de Química Orgánica de Síntesis, Facultad de Ciencias Bioquímicas y Farmacéuticas, 2000 Rosario, Santa Fé, Argentine

Correspondence e-mail: srussi@fq.edu.uy

#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.051 wR factor = 0.167 Data-to-parameter ratio = 18.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved

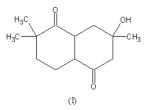
# 7-Hydroxy-2,2,7-trimethylperhydronaphthalene-1,5-dione

The title compound,  $C_{13}H_{20}O_3$ , crystallizes as a racemic mixture in which the two enantiomeric forms of the molecule are related by an inversion center. The crystal structure confirms the one proposed on the basis of spectroscopic data and provides the information required to assign the relative stereochemistry of the chiral centers of the molecule. The crystal packing is directed by intermolecular hydrogen bonds and short intramolecular C-H···O contacts.

Received 13 July 2004 Accepted 15 July 2004 Online 31 July 2004

# Comment

Decalin ring systems represent a common structural motif of a large variety of active molecules (Fraga, 2001; Faulkner, 2002; Hanson, 2003). In particular, terpenoids and steroids display a wide range of biological activities. The preparation of this bicyclic unit with appropriate stereochemistry has attracted widespread attention in the field of medicinal and synthetic organic chemistry, not only in consideration of their abundance in nature, but also their usefulness as versatile synthetic building blocks. Consequently, numerous methodologies have been developed and continue to be explored for its synthesis (Varner & Grossman, 1999; Tokoroyama, 2000). Among them, annulations routes based on the combination of classical and modified carbonyl reactions have proved to be very useful strategies to access various carbocyclic frameworks.



Recently, in the course of our studies on the application of cyclocondensation reactions of acyclic triketones, a racemic mixture of the title compound, (I), was prepared (Mischne, 2003). The structure was elucidated on the basis of the analysis of its <sup>1</sup>H and <sup>13</sup>C NMR spectroscopic data, and a preliminary relative stereochemical assignment was made on the basis of <sup>1</sup>H–<sup>1</sup>H *J*-coupling values (Mischne, 2003). However, due to the potential use of this compound as an intermediate in the synthesis of more complex natural product analog products, it was important to unambiguously confirm the proposed structure and the relative stereochemistry by single-crystal X-ray diffraction.

The compound crystallizes in the triclinic space group  $P\overline{1}$  with an inversion center relating the two enantiomeric forms of the molecule. Fig. 1 shows one of these enantiomeric forms,

1951 reflections with  $I > 2\sigma(I)$ 

 $R_{\rm int} = 0.030$ 

 $\theta_{\rm max} = 27.5^\circ$ 

 $h = -7 \rightarrow 7$ 

 $k = -2 \rightarrow 12$ 

 $l = -14 \rightarrow 14$ 

3 standard reflections

every 150 reflections

intensity decay: 2.5%

 $w = 1/[\sigma^2(F_o^2) + (0.0851P)^2]$ 

where  $P = (F_0^2 + 2F_c^2)/3$ 

Extinction correction: SHELXL97

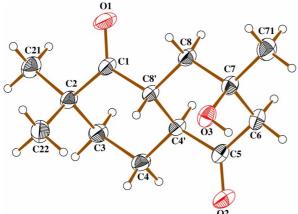
Extinction coefficient: 0.055 (11)

+ 0.1359P]

 $\Delta \rho_{\rm max} = 0.39 \ {\rm e} \ {\rm \AA}^{-3}$ 

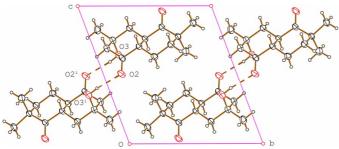
 $\Delta \rho_{\rm min} = -0.20 \text{ e } \text{\AA}^{-3}$ 

 $(\Delta/\sigma)_{\rm max} < 0.001$ 



## Figure 1

ZORTEP drawing (Zsolnai & Pritzkow, 1995) of (I), showing the atomlabeling scheme. Displacement ellipsoids are drawn at the 30% probability level.



#### Figure 2

ZORTEP packing diagram (Zsolnai & Pritzkow, 1995) projected on to the bc plane. Intermolecular hydrogen bonds are marked as dashed lines. [Symmetry code: (i) 2 - x, 2 - y, 1 - z.]

consisting of two trans-fused cyclohexanone rings substituted by one hydroxyl and three methyl groups. All the geometric parameters (Table 1) are consistent with those reported for other trans-decalindione systems (Ariel & Trotter, 1984; Garcia et al., 1991).

As in the case of the trans-decalin-1,5-dione structure, each cyclohexanone ring has an almost ideal chair conformation. The puckering parameters are  $\theta = 0.66 (19)^{\circ}$  and  $Q_T =$ 0.5615 (19) Å for the C1–C8' ring and  $\theta = 2.02$  (19)° and  $Q_T =$ 0.5735 (19) Å for the C4'-C8' ring (Cremer & Pople, 1975). The  $sp^2$  hybridization of atoms C1 and C5 place the carbonyl O atoms (O1 and O3) coplanar with the C atoms of the rings in which they are located and bisectional to the Cremer and Pople normal planes. The respective angles with these planes are 53.29 (11) and 54.53 (11)°. The hydroxyl and methyl (C22) groups on C7 and C2 adopt axial positions, the other two methyl substituent groups (C21 and C71) being in equatorial positions.

The molecule has three chiral centers, viz. C4', C7 and C8'. Since the compound was obtained and crystallized as a racemic mixture, only a relative assignment of the absolute configuration of these chiral centers can be made: an S(R)configuration for C7 and C8' and R(S) configuration for C4'.

The crystal packing is directed by a pair of intermolecular hydrogen bonds that link two inversion-related molecules, generating discrete dimeric units  $[O3-H3\cdots O2^{i}: O3-H3 =$   $0.82 \text{ Å}, \text{H3} \cdots \text{O2} = 2.09 \text{ Å}, \text{O3} - \text{H3} \cdots \text{O2} = 172^{\circ}; \text{ symmetry}$ code: (i) 2 - x, 2 - y, 1 - z]. Another short intramolecular contact involving atom O3 (C8'-H8'···O3: C8'-H8' =  $0.98 \text{ Å}, \text{H8}' \cdots \text{O3} = 2.50 \text{ Å} \text{ and } \text{C8}' - \text{H8}' \cdots \text{O3} = 101^{\circ}$ ) also influences the packing and stabilizes the conformation of the molecule.

## **Experimental**

The title compound was synthesized via the sequential intramolecular Michael/aldol strategy, as described previously (Mischne, 2003). Suitable single crystals for X-ray diffraction were obtained by vapor diffusion (ethyl acetate/petroleum ether) at room temperature.

Crystal data

$C_{13}H_{20}O_{3}$	Z = 2
$M_r = 224.29$	$D_x = 1.235 \text{ Mg m}^{-3}$
Triclinic, $P\overline{1}$	Mo $K\alpha$ radiation
a = 6.0562 (13)  Å	Cell parameters from 25
b = 9.827 (2)  Å	reflections
c = 11.1075 (16)  Å	$\theta = 35.0-47.5^{\circ}$
$\alpha = 109.289 \ (14)^{\circ}$	$\mu = 0.09 \text{ mm}^{-1}$
$\beta = 102.766 \ (15)^{\circ}$	T = 293 (2)  K
$\gamma = 92.771 \ (19)^{\circ}$	Block, colorless
$V = 603.2 (2) \text{ Å}^3$	$0.30 \times 0.15 \times 0.10 \ \text{mm}$

# Data collection

Rigaku AFC-7S diffractometer  $\theta/2\theta$  scans Absorption correction:  $\psi$  scan (MSC/AFC Diffractometer Control Software; Molecular Structure Corporation, 1993)  $T_{\min} = 0.965, \ T_{\max} = 0.991$ 3565 measured reflections

2768 independent reflections

### Refinement

Refinement on  $F^2$  $R[F^2 > 2\sigma(F^2)] = 0.051$  $wR(F^2) = 0.167$ S = 1.042768 reflections 149 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

C1-O1	1.2115 (19)	C2-C22	1.533 (3)
C1-C8′	1.5222 (19)	C2-C3	1.540 (2)
C1-C2	1.528 (2)	C6-C5	1.500 (2)
C8-C8′	1.522 (2)	C6-C7	1.542 (2)
C8-C7	1.528 (2)	O2-C5	1.214 (2)
C4′-C4	1.514 (2)	C7-O3	1.429 (2)
C4′-C5	1.515 (2)	C7-C71	1.522 (2)
C4′-C8′	1.551 (2)	C4-C3	1.524 (2)
C2-C21	1.528 (2)		
O1-C1-C8′	121.32 (14)	C1-C2-C3	107.03 (13)
O1-C1-C2	122.12 (14)	C5-C6-C7	110.42 (13)
C8' - C1 - C2	116.48 (12)	O3-C7-C71	109.78 (14)
C8′-C8-C7	111.47 (12)	C8-C7-C6	109.90 (13)
C4-C4'-C8'	112.03 (13)	C4′-C4-C3	110.79 (12)
C5-C4'-C8'	107.69 (12)	C4-C3-C2	113.71 (13)
C8-C8'-C1	112.79 (12)	O2-C5-C6	122.73 (14)
C1-C8'-C4'	108.38 (12)	O2-C5-C4′	122.31 (15)
C21-C2-C22	108.60 (15)	C6-C5-C4′	114.95 (13)

The H atoms were included in geometrically calculated positions, with fixed C–H distances of 0.97 Å for CH<sub>2</sub> groups and 0.96 Å for CH<sub>3</sub> groups, and with an O–H distance of 0.82 Å for the hydroxyl group. The  $U_{\rm iso}({\rm H})$  values for these H atoms were fixed at 1.2 $U_{\rm eq}$  of the attached atom.

Data collection: (*MSC/AFC Diffractometer Control Software*; Molecular Structure Corporation, 1993); cell refinement: *MSC/AFC Diffractometer Control Software*; data reduction: *MSC/AFC Diffractometer Control Software*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ZORTEP* (Zsolnai & Pritzkow, 1995); software used to prepare material for publication: *PLATON* (Spek, 2003) and *encIFer* (Allent *et al.*, 2004).

The authors thank the Spanish Research Council (CSIC) for the free-of-charge license to the Cambridge Structural Database (Allen, 2002).

## References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Allen, F. H., Johnson, O., Shields, G. P., Smith, B. R. & Towler, M. (2004). J. Appl. Cryst. 37, 335–338.
- Ariel, S. & Trotter, J. (1984). Acta Cryst. C40, 1253-1255.
- Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
- Faulkner, D. (2002). Nat. Prod. Rep. 19, 1-48.
- Fraga, B. M. (2001). Nat. Prod. Rep. 19, 650-672.
- Garcia, J. G., Fronczek, F. R. & McLaughlin, M. L. (1991). Acta Cryst. C47, 204–206.
- Hanson, R. J. (2003). Nat. Prod. Rep. 20, 70-78.
- Mischne, M. (2003). Tetrahedron Lett. 44, 5823-5826.
- Molecular Structure Corporation (1993). MSC/AFC Diffractometer Control Software. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (2003). PLATON. Utrecht University, The Netherlands.
- Tokoroyama, T. (2000). Synthesis, pp. 611-633.
- Varner, M. A. & Grossman, R. B (1999). Tetrahedron, 55, 13867-13886.
- Zsolnai, L. & Pritzkow, H. (1995). ZORTEP. University of Heidelberg, Germany.